Acta Crystallographica Section E

## Structure Reports

 OnlineISSN 1600-5368

Adrienne F. M. Fleming, ${ }^{\text {a }}$ Fintan Kelleher, ${ }^{\text {a }}$ Mary F. Mahon, ${ }^{\text {b }}$ * John McGinley, ${ }^{\text {a }}$ Kieran C. Molloy ${ }^{\text {b }}$ and Vipa Prajapati ${ }^{\text {a }}$
${ }^{\text {a }}$ Department of Applied Science and Advanced Smart Materials Research Center, Institute of Technology, Tallaght, Dublin 24, Ireland, and ${ }^{\mathbf{b}}$ Department of Chemistry, University of Bath, Claverton Down, Bath BA1 7AY, England

Correspondence e-mail: m.f.mahon@bath.ac.uk

## Key indicators

Single-crystal X-ray study
$T=150 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.004 \AA$
$R$ factor $=0.048$
$w R$ factor $=0.131$
Data-to-parameter ratio $=23.1$
For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

[^0]
# 1,4-Bis[2-(6-bromohexyl)-2H-tetrazol-5-yl]benzene 

The 150 K structure of the title compound, $\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{Br}_{2} \mathrm{~N}_{8}$, has been shown to exhibit liquid-crystal alignment in the gross array, enhanced by the presence of intermolecular $\mathrm{Br} \cdots \mathrm{Br}$ interactions. The asymmetric unit consists of one-half of a molecule, the remainder being generated via a crystallographic inversion centre located at the centre of the benzene ring.

## Comment

The synthesis of tetrazoles from the cycloaddition reaction between an azide and a nitrile is well established (Butler, 1996). Regioselective alkylation of tetrazoles has been the subject of several investigations during the last 20 years (Bethel et al., 1999; Goodger et al., 1996; Hill et al., 1996; Zubarev \& Ostrovskii, 2000). Our group has recently studied the alkylation of 1,4 -bis( 1 H -tetrazol-5-yl)benzene, (I), with various alkyl halides to give bifunctional products of type (II) with N-2 substitution in both tetrazole rings (Fleming et al., 2004). These compounds are intermediates in the generation of tetratetrazole macrocycles of general formula (III) which include a cavity of variable dimensions tailored by both the length and flexibility of the bridging groups $X$ and $Y$ (Butler \& NíBhrádaigh, 1994; Butler et al., 1992, 2001; Butler \& Fleming, 1997). As part of our ongoing studies on this family of compounds, the structure of 1,4-bis[2-(6-bromohexyl)-2 H -tetrazol-5-yl]benzene, (II), is now reported (Fig. 1).

(I)

(III)

(II)

The asymmetric unit consists of one half of a molecule, the remainder being generated via a crystallographic inversion centre located at the centre of the benzene ring. Both tetrazole rings are coplanar with the benzene ring to which they are attached (the largest deviation from the least-squares plane is $0.023 \AA$ for atom C4). Analysis of the gross structure reveals slipped $\pi$ stacking, with a distance of $3.38 \AA$ between the leastsquares planes of the tetrazole ring and the benzene ring of its closest neighbour (Fig. 2). It is also probable that liquid crystal

Received 9 November 2004 Accepted 16 November 2004 Online 20 November 2004


Figure 1
Plot of the title molecule. Displacement ellipsoids are drawn at the $30 \%$ probability level. Unlabelled atoms are related to labelled atoms by the symmetry operator $(-x,-y, 1-z)$.


Figure 2
Plot of two neighbouring molecules in the crystal structure of (II), showing $\pi$ stacking.


Figure 3
Partial packing diagram for (II), illustrating intermolecular interactions and liquid-crystal alignment.
alignment is present along the $c$ axis in this structure, evidenced by an intermolecular distance of 3.4802 (4) $\AA$ between proximate terminal Br atoms (Fig. 3). This distance is considerably shorter than twice the bromine van der Waals radius ( $3.90 \AA$ ) and is within the range of those values previously reported, 3.415-3.691 $\AA$ (Christofi et al., 2000; Kuhn et al., 2004; Ruthe et al., 1997; Savinsky et al., 2001).

## Experimental

A suspension of 1,4-bis( 1 H -tetrazol-5-yl)benzene ( $1.0 \mathrm{~g}, 4.7 \mathrm{mmol}$ ), methanol ( 30 ml ) and triethylamine ( $1.90 \mathrm{~g}, 19 \mathrm{mmol}$ ) was stirred at 333 K for an hour. 1,6-Dibromohexane ( $3.6 \mathrm{~g}, 14 \mathrm{mmol}$ ) was then added and the mixture was heated under reflux for 5 h . The solvent was removed and the product was purified using column chromatography (hexane/ethyl acetate $80 / 20$ initially followed by $60 / 40$ ). Crystals suitable for X-ray diffraction were grown from acetonitrile.

## Crystal data

| $\mathrm{C}_{20} \mathrm{H}_{28} \mathrm{Br}_{2} \mathrm{~N}_{8}$ | $Z=1$ |
| :--- | :--- |
| $M_{r}=540.32$ | $D_{x}=1.604 \mathrm{Mg} \mathrm{m}^{-3}$ |
| Triclinic, $P \overline{1}$ | Mo $K \alpha$ radiation |
| $a=4.6290(1) \AA$ | Cell parameters from 20025 |
| $b=5.7030(1) \AA$ | reflections |
| $c=21.7111(6) \AA$ | $\theta=12-18^{\circ}$ |
| $\alpha=87.494(1)^{\circ}$ | $\mu=3.65 \mathrm{~mm}^{-1}$ |
| $\beta=85.631(1)^{\circ}$ | $T=150(2) \mathrm{K}$ |
| $\gamma=78.312(2)^{\circ}$ | Irregular tablet, colourless |
| $V=559.40(2) \AA^{\circ}$ | $0.60 \times 0.40 \times 0.12 \mathrm{~mm}$ |

## Data collection

Nonius KappaCCD diffractometer $\varphi$ and $\omega$ scans
Absorption correction: multi-scan (Blessing, 1995)

$$
T_{\min }=0.187, T_{\max }=0.65
$$

8752 measured reflections
3144 independent reflections

## Refinement

Refinement on $F^{2}$

$$
\begin{aligned}
& w=1 /\left[\sigma^{2}\left(F_{o}{ }^{2}\right)+(0.0798 P)^{2}\right. \\
& \quad+0.1247 P] \\
& \text { where } P=\left(F_{o}{ }^{2}+2 F_{c}^{2}\right) / 3 \\
& (\Delta / \sigma)_{\max }=0.001 \\
& \Delta \rho_{\max }=1.11 \mathrm{e}^{-3} \AA^{-3} \\
& \Delta \rho_{\min }=-1.18 \mathrm{e} \AA^{-3}
\end{aligned}
$$

$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.048$
$w R\left(F^{2}\right)=0.131$
$S=1.05$
3144 reflections
136 parameters
2680 reflections with $I>2 \sigma(I)$
$R_{\text {int }}=0.075$
$\theta_{\text {max }}=29.9^{\circ}$
$h=-6 \rightarrow 6$
$k=-7 \rightarrow 7$
$l=-30 \rightarrow 30$

H -atom parameters constrained

H atoms were included at calculated positions and constrained to an ideal geometry, with C-H distances of $0.98 \AA$ and with $U_{\text {iso }}(\mathrm{H})=$ $1.2 U_{\text {eq }}(\mathrm{C})$. The largest peak and deepest hole in the final difference map are located 849 and $0.769 \AA$, respectively, from atom Br 1 .

Data collection: COLLECT (Hooft, 1998); cell refinement: HKL DENZO (Otwinowski \& Minor, 1997); data reduction: DENZO and SCALEPACK (Otwinowski \& Minor, 1997); program(s) used to solve structure: SHELXS97 (Sheldrick, 1990); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEX (McArdle, 1995); software used to prepare material for publication: SHELXL97.

## References

Bethel, P. A., Hill, M. S., Mahon, M. F. \& Molloy, K. C. (1999). J. Chem. Soc. Perkin Trans. 1, pp. 3507-3514.
Blessing, R. H. (1995). Acta Cryst. A51, 33-37.
Butler, R. N. (1996). Comprehensive Heterocyclic Chemistry, Vol. 4, edited by A. R. Katrilzky, C. W. Rees \& E. F. V. Scriven. Oxford: Pergamon.

Butler, R. N. \& Fleming, A. F. M. (1997). J. Heterocycl. Chem. 34, 691-693.
Butler, R. N., McGinley, J., Mahon, M. F., Molloy, K. C. \& NíBhrádaigh, E. P. (2001) Acta Cryst. E57, o195-o197.

Butler, R. N. \& NíBhrádaigh, E. P. (1994). J. Chem. Res. (S), pp. 148-149.
Butler, R. N., Quinn, K. F. \& Welke, B. (1992). J. Chem. Soc. Chem. Commun. pp. 1481-1482.
Christofi, A. M., Garratt, P. J., Hogarth, G. \& Steed, J. W. (2000). J. Chem. Soc. Dalton Trans. pp. 2137-2144.
Fleming, A. F. M., Kelleher, F., Mahon, M. F., McGinley, J. \& Prajapati, V. (2004). Unpublished work.

Goodger, A., Hill, M. S., Mahon, M. F., McGinley, J. \& Molloy, K. C. (1996). J. Chem. Soc. Dalton Trans. pp. 847-852.
Hill, M. S., Mahon, M. F., McGinley, J. \& Molloy, K. C. (1996). J. Chem. Soc. Dalton Trans. pp. 835-845.
Hooft, R. (1998). COLLECT. Nonius BV, Delft, The Netherlands.
Kuhn, N., Abu-Rayyan, A., Eichele, K., Schwarz, S. \& Steimann, M. (2004). Inorg. Chim. Acta, 357, 1799-1804.
McArdle, P. (1995). J. Appl. Cryst. 28, 65.
Otwinowski, Z. \& Minor, W. (1997). Methods in Enzymology, Vol. 276, Macromolecular Crystallography, Part A, edited by C. W. Carter Jr and R. M. Sweet, pp. 307-326. New York: Academic Press.

Ruthe, F., Du Mont, W.-W. \& Jones, P. G. (1997). Chem. Commun. p. 1947.
Savinsky, R., Hopf, H., Dix, I. \& Jones, P. G. (2001). Eur. J. Org. Chem. pp. 4595-4606.
Sheldrick, G. M. (1990). Acta Cryst. A46, 467-473.
Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
Zubarev, V. Y. \& Ostrovskii, V. A. (2000). Chem. Heterocycl. Compds, 53, 1421-1448.


[^0]:    (C) 2004 International Union of Crystallography Printed in Great Britain - all rights reserved

